

Novel Synthesis of (+)-4-Acetoxy-3-hydroxyethyl-2-azetidinone from Carbohydrate

A Formal Total Synthesis of (+)-Thienamycin[†]

Sir:

(+)-4-Acetoxy-3-hydroxyethyl-2-azetidinone, that is (3*R*,4*R*)-4-acetoxy-3-[(*R*)-1'-hydroxyethyl]-2-azetidinone (**1**), and its derivatives have been well-known as the highly versatile intermediates¹⁾ for the synthesis of carbapenem antibiotics such as thienamycin (**2**)²⁾, imipenem, meropenem³⁾ and so on (Fig. 1).

The synthesis of **1** was initiated by Sankyo group⁴⁾, followed by Merck group⁵⁾, and culminated in the practical preparation by two Japanese companies^{6,7)} using Noyori-Murahashi's asymmetric procedures and chem-enzymatic procedures, respectively.

(+)-Thienamycin (**2**) was discovered in fermentation broths of *Streptomyces cattleya* to show exceptional antibacterial potency and spectrum²⁾. The first stereocontrolled synthesis of **2** has been reported by Merck group⁸⁾, and the transformation of **1** to (+)-thienamycin (**2**) was also made more attractive by another Merck group⁹⁾. Consequently, the synthesis of (+)-4-acetoxy-3-hydroxyethyl-2-azetidinone (**1**) constitutes a formal total synthesis of (+)-thienamycin (**2**).

Herein, we report a novel enantiospecific synthesis of **1** from a carbohydrate through the skeletal rearrangement and stereoselective epimerization. Our starting material is commercially available methyl 2-amino-2,6-dideoxy- α -D-glucopyranoside (**3**), which has been also isolated from natural sources¹⁰⁾.

Reaction of **3** with *o*-benzenedisulfonyl dichloride¹¹⁾ gave the cyclic sulfonate **4** (Table 1), which was submitted to our developed skeletal rearrangement¹¹⁾ including ring-contraction with potassium *tert*-butoxide (Scheme 1). The resulting 3-formyl-furanoside **5** was oxidized to the mixture of the carboxylic acids, which were readily separated on silica gel column chromatography (CHCl₃-MeOH 8:1). Both carboxylic acids were purified on Dowex 50 \times 8 resin (Na-type) with H₂O to give the sodium salt **6** and its C-3 epimer **7** in 49% and 42% yields in 2 steps from **4**, respectively. Practically, both compounds could be used for the synthesis without separation, because they were found to be efficiently converted to a single lactone **11** by

stereoselective epimerization later on. However, we now describe the synthesis from each of both compounds **6** and **7** as follows.

Removal of the *N*-sulfonyl group of **6** by Birch reduction produced the corresponding amino acid **8**. This was hydrolyzed and then esterified to give the furanose **9**. The oxidation of **9** to the lactone **10** was the key step of our strategy, although the lactone could not be obtained under usual oxidation conditions. Finally, we found that, on exposure to Ag₂CO₃/Celite¹²⁾ in benzene, the furanose **9** was smoothly oxidized to the γ -lactone **10** in spite of the presence of the amino group.

The next important operation in the synthesis was to epimerize stereoselectively the configurations at C-2 and C-3 positions of **10**. After a variety of conditions were examined, the best result was realized by using DBU in MeOH to afford predominantly the desired amino ester **11**.

Similarly, the epimer **7** was transformed to **11** through **12** and **13** in 57% overall yield.

The structure of **11** was reasonably confirmed by the NMR studies of **10**, **11** and **13**.

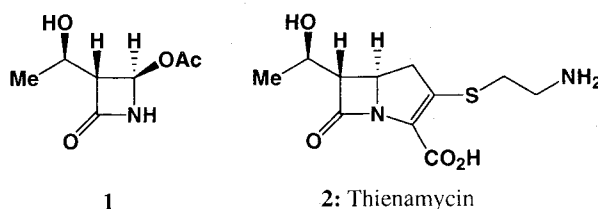
These results indicated that the C-4 configuration of **10** or **13** controlled the stereoselective construction of the C-2 and C-3 configurations of **11**.

Hydrolysis of **11** with 2M NaOH according to the reported procedures¹³⁾ led to the hydroxy acid **14**, which was in turn submitted to the β -lactam formation (Scheme 2).

For our purpose, a Grignard-mediated cyclization of the silylated derivative seemed most promising¹⁴⁾. Thus, **14** was silylated with trimethylsilyl chloride and hexamethyldisilazane (HMDS), and followed by treatment with *tert*-butylmagnesium chloride to give the bis-silylated β -lactam **15**¹³⁾.

Oxidative decarboxylation⁹⁾ of **15** by Pb(OAc)₄ gave

Fig. 1.



[†] This paper is dedicated to the memory of Sir EDWARD ABRAHAM, a pioneer of β -lactam antibiotics science.

Table 1. Physico-chemical properties of compounds.

Compds.	Mp (°C)	$[\alpha]_D$	$^1\text{H-NMR}$ (600 or 300MHz; δ ppm; J Hz)
1	108-110	+63° (<i>c</i> 0.74, CHCl_3)	(CDCl_3): δ 1.35(3H, d, $J=6.2$), 2.13(3H, s), 3.23(1H, dd, $J=1.0\&6.2$), 4.22(1H, dq, $J=6.2\&6.2$), 5.81(1H, d, $J=1.0$), 6.58(1H, br s)
4	190-196 (decomp.)	+142° (<i>c</i> 1.10, Me_2CO)	(CDCl_3): δ 1.36(3H, d, $J=6.2$), 2.73(1H, br s), 3.44(3H, s), 3.46(1H, dd, $J=9.4\&9.4$), 3.78(1H, dq, $J=6.2\&9.4$), 3.91(1H, ddd, $J=3.6, 10.4\&10.4$), 4.82(1H, d, $J=3.6$), 5.03(1H, dd, $J=9.4\&10.4$), 5.61(1H, br d, $J=10.4$), 7.70(1H, ddd, $J=1.4, 8.0\&8.0$), 7.80(1H, ddd, $J=1.4, 8.0\&8.0$), 8.24(1H, dd, $J=1.4\&8.0$), 8.25(1H, dd, $J=1.4\&8.0$)
6	Wax	-7.6° (<i>c</i> 0.71, MeOH)	(CD_3OD): δ 1.15(3H, d, $J=6.4$), 2.87(1H, dd, $J=8.4\&8.4$), 2.89(3H, s), 4.24(1H, d, $J=4.8$), 4.35(1H, dq, $J=6.4\&8.4$), 4.36(1H, dd, $J=4.8\&8.4$), 7.61(1H, ddd, $J=1.4, 8.0\&8.0$), 7.68(1H, ddd, $J=1.4, 8.0\&8.0$), 8.15(1H, dd, $J=1.4\&8.0$), 8.24(1H, dd, $J=1.4\&8.0$)
7	Wax	-2.5° (<i>c</i> 0.71, MeOH)	(CD_3OD): δ 1.27(3H, d, $J=6.2$), 2.65(1H, dd, $J=7.6\&11.0$), 2.68(3H, s), 4.18(1H, dd, $J=5.0\&11.0$), 4.24(1H, dq, $J=6.2\&7.6$), 4.25(1H, d, $J=5.0$), 7.61(1H, ddd, $J=1.4, 8.0\&8.0$), 7.67(1H, ddd, $J=1.4, 8.0\&8.0$), 8.11(1H, dd, $J=1.4\&8.0$), 8.25(1H, dd, $J=1.4\&8.0$)
8•HCl	210-215 (decomp.)	+182° (<i>c</i> 1.18, MeOH)	($\text{CD}_3\text{OD-DCl/D}_2\text{O}$): δ 1.24(3H, d, $J=6.2$), 3.32(1H, dd, $J=7.6\&8.4$), 3.44(3H, s), 4.09(1H, dd, $J=4.8\&7.6$), 4.63(1H, dq, $J=6.2\&8.4$), 5.14(1H, d, $J=4.8$)
9•HCl	Oil		α -Anomer ($\text{DCl/D}_2\text{O}$): δ 1.18(3H, d, $J=6.4$), 3.53(1H, dd, $J=7.8\&8.4$), 3.81(3H, s), 4.25(1H, dd, $J=5.0\&7.8$), 4.81(1H, dq, $J=6.4\&8.4$), 5.69(1H, d, $J=5.0$)
10•HCl	152-157	+102° (<i>c</i> 1.06, MeOH)	($\text{CD}_3\text{OD-DCl/D}_2\text{O}$): δ 1.35(3H, d, $J=6.8$), 3.85(3H, s), 3.99(1H, dd, $J=9.0\&11.2$), 4.81(1H, d, $J=11.2$), 5.15(1H, dq, $J=6.8\&9.0$)
11	Oil	-26° (<i>c</i> 1.12, CHCl_3)	(CDCl_3): δ 1.53(3H, d, $J=6.0$), 2.83(1H, dd, $J=9.6\&11.2$), 3.81(3H, s), 4.09(1H, d, $J=11.2$), 4.53(1H, dq, $J=6.0\&9.6$)
12•HCl	202-206 (decomp.)	+85° (<i>c</i> 0.32, MeOH)	($\text{CD}_3\text{OD-DCl/D}_2\text{O}$): δ 1.39(3H, d, $J=6.2$), 2.99(1H, dd, $J=7.8\&10.0$), 3.44(3H, s), 4.03(1H, dd, $J=5.2\&10.0$), 4.42(1H, dq, $J=6.2\&7.8$), 5.15(1H, d, $J=5.2$)
13•HCl	150-154	-14° (<i>c</i> 1.32, MeOH)	($\text{CD}_3\text{OD-DCl/D}_2\text{O}$): δ 1.54(3H, d, $J=6.6$), 3.65(1H, dd, $J=2.4\&8.6$), 3.84(3H, s), 4.80(1H, d, $J=8.6$), 5.03(1H, dq, $J=2.4\&6.6$)
14	171-174 (decomp.)	+12° (<i>c</i> 0.14, MeOH)	($\text{DMSO}-d_6$): δ 1.13(3H, d, $J=6.2$), 2.91(1H, dd, $J=3.8\&6.2$), 3.54(1H, d, $J=3.8$), 3.57(3H, s), 4.04(1H, dq, $J=6.2\&6.2$)
15	Wax	-33° (<i>c</i> 0.17, CHCl_3)	(CDCl_3): δ 0.12(9H, s), 0.28(9H, s), 1.24(3H, d, $J=12.0$), 3.31(1H, dd, $J=6.0\&6.0$), 4.21(1H, d, $J=6.0$), 4.28(1H, dq, $J=6.0\&12.0$)

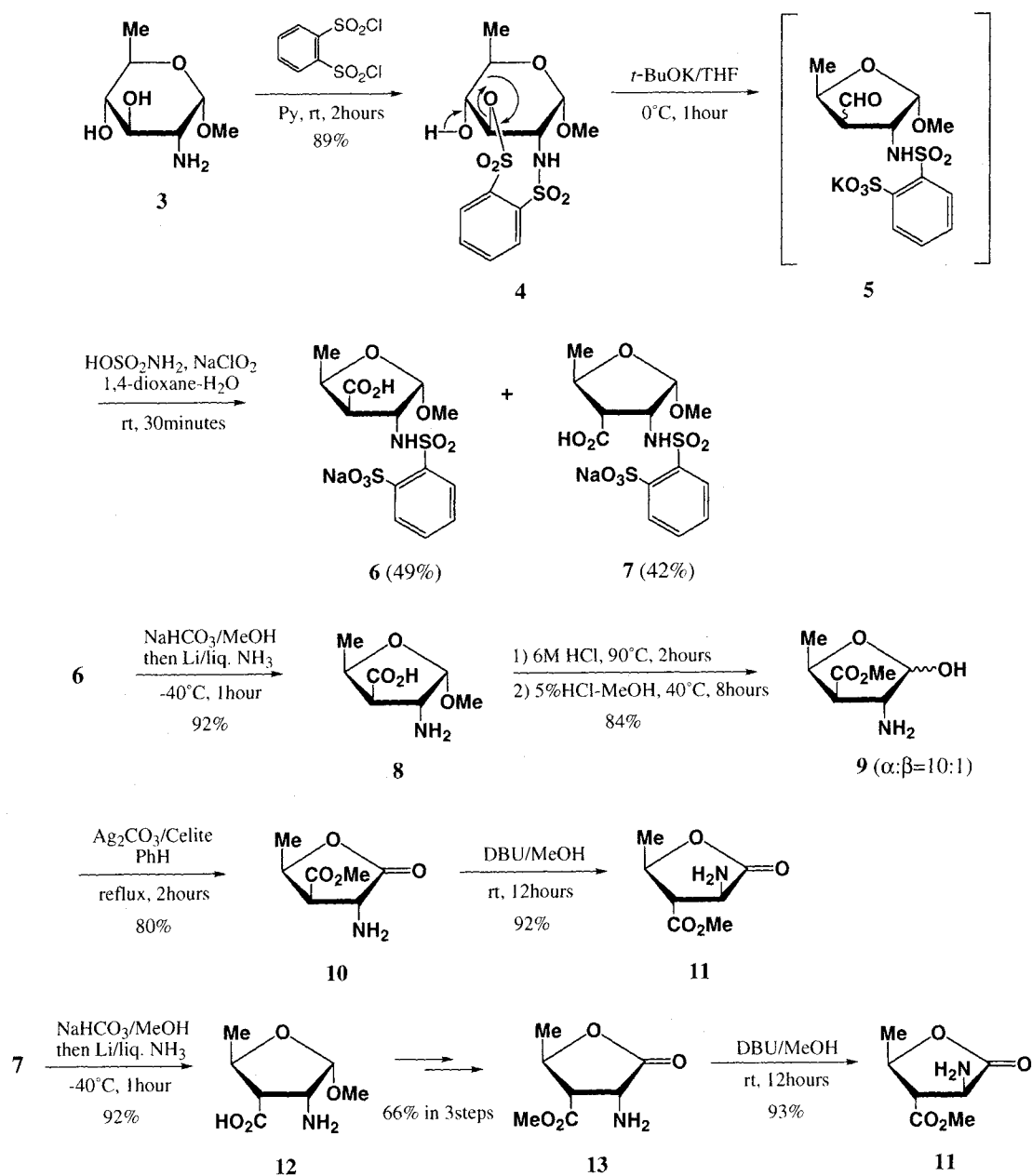
exclusively the desired (+)-4-acetoxy-3-hydroxyethyl-2-azetidinone (**1**) with removal of silyl groups. This was identical in all respects with the authentic sample^{9,13}, which was prepared from commercially available *O*-tert-butylidimethylsilyl derivative.

Overall, the yield was approximately 32% in 12 steps from **3**. Key steps include our developed skeletal rearrangement with ring-contraction, oxidation of the 2-aminofuranose, and stereoselective epimerization to the desired configurations.

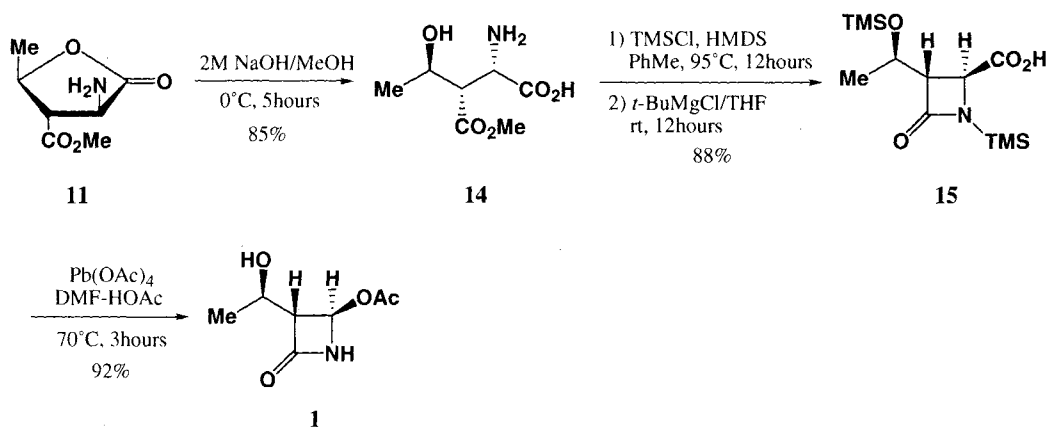
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Scheme 1.



Scheme 2.



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